Synthesis and Structural and Electrochemical Properties of $Li[Co_{1\text{-}2x}(Li_{1/3}Mn_{2/3})_x(Ni_{1/2}Mn_{1/2})_x]O_2$

Li[Co_{1-2x}(Li_{1/3}Mn_{2/3})_x(Ni_{1/2}Mn_{1/2})_x]O₂의 합성 및 전기적, 구조적 특성에 관한 연구

<u>Ki Soo Park</u>, Sung Jang Jin, and Kee Suk Nahm School of Chemical Engineering and Technology, ChonbukNationalUniversity

An advanced materials of $\text{Li}[\text{Co}_{1\text{-}2x}(\text{Li}_{1/3}\text{Mn}_{2/3})_y(\text{Ni}_{1/2}\text{Mn}_{1/2})_y]O_2$ was developed and studied their properties of two compositions of solid solutions with layered manganese oxides. The materials were satisfied the theoretical manganese oxidation state of 4+ over that equation.

In order to confirm manganese state of 4+ in Li[Co_{1-2x}(Li_{1/3}Mn_{2/3})_y(Ni_{1/2} Li[Li_{1/15}Ni_{1/10}Co_{3/5}Mn_{7/30}]O₂ (x=0.2)two compositions, $Mn_{1/2})_{v}]O_{2}$ $\text{Li}[\text{Li}_{2/15}\text{Ni}_{1/5}\text{Co}_{1/5}\text{Mn}_{7/15}]\text{O}_2$ (x=0.4) were synthesized using a sol-gel method. $\label{eq:Li_Li_1/15} Li[Li_{1/15}Ni_{1/10}Co_{3/5}Mn_{7/30}]O_2 \quad (x=0.2) \quad \text{and} \quad Li[Li_{2/15}Ni_{1/5}Co_{1/5}Mn_{7/15}]O_2 \quad (x=0.4) \quad were$ indexed typical layered structure (S. G: R 3m). The initial discharge capacities of samples $\text{Li}[\text{Li}_{1/15}\text{Ni}_{1/10}\text{Co}_{3/5}\text{Mn}_{7/30}]O_2$ (x=0.2) and $\text{Li}[\text{Li}_{2/15}\text{Ni}_{1/5}\text{Co}_{1/5}\text{Mn}_{7/15}]O_2$ (x=0.2) = 0.4) were 178 and 185 mAh/g, respectively. The subsequent capacities of samples showed 173 and 190 mAh/g after 20 cycles. The capacity retentions (initial capacity / terminal capacity) of samples Li[Li_{1/15}Ni_{1/10}Co_{3/5}Mn_{7/30}]O₂ (x= and $\text{Li}[\text{Li}_{2/15}\text{Ni}_{1/5}\text{Co}_{1/5}\text{Mn}_{7/15}]\text{O}_2$ (x=0.4) showed 97.2 and respectively. Li ion on the transition metal layers delivered high discharge capacities and retention ratios.