Covalent attachment of streptavidin on colloidal gold nanoparticle and its characterization

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The synthesis and characterization of biomolecule-conjugated nanoparticles is currently a very active research field 1,2). There were many works about the conjugation of biomolecule to nanoparticles as noncovalent 3. However, little has been reported on the covalent conjugation of biomolecule to nanoparticles. In this study, we present a method for the covalent attachment of streptavidin onto alkanethiol-modified gold surfaces by forming amide bonds via an N-hydroxysulfosuccinimide (NHSS) ester intermediate. The colloidal gold nanoparticle of 13 nm diameter were prepared by the citrate reduction of HAuCl₄³⁾. The solution of colloidal particle was characterized by an adsorption maximum at 520 nm. Transmission electron microscopy (TEM) indicated a particle size of 13 nm. The gold colloids were immobilized with an organic molecule containing thiol group such as 11-mercaptoundecanoic acid (MUA), and mercaptosuccinic acid (MSA) as covalent. The amide bond is formed in two steps: the terminal carboxylic acid groups of an alkanethiol self-assembled monolayer (SAM) are first activated to the N-hydroxysulfosuccinimide (NHSS) ester, followed by reaction of this MUA-NHSS ester with amino groups of streptavidin. The obtained streptavidin conjugated gold nanoparticles were characterized by FT-IR, transmission electron microscopy (TEM), and UV spectroscopy. With the streptavidin conjugated nanoparticles, we are developing the universal nanoparticle-based labeling process by the use of the biotin-streptavidin coupling for the binding of biotinylated target DNA to streptavidin-conjugated gold colloids.

Reference

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