

## BFB11

Characterization of structural and electrochemical properties of  $\text{LiMn}_2\text{O}_4$  coated with  $\text{LiNi}_{1-x}\text{Co}_x\text{O}_2$  ( $x = 0.2$  and  $1$ ) using Solution-based Chemical Process

$\text{LiNi}_{1-x}\text{Co}_x\text{O}_2$  ( $x = 0.2$  and  $1$ )이 표면 코팅된  $\text{LiMn}_2\text{O}_4$ 의 전기화학적 특성 및 구조적 특성에 관한 연구

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The surface coating of  $\text{LiMn}_2\text{O}_4$  using a gel precursor of  $\text{LiNi}_{1-x}\text{Co}_x\text{O}_2$  ( $x = 0.2$  and  $1$ ) from a chemical process was attempted in order to enhance the cycle stability of  $\text{LiMn}_2\text{O}_4$  at elevated temperature. When the surface of  $\text{LiMn}_2\text{O}_4$  was coated with  $\text{LiCoO}_2$  coating solution, the surface of  $\text{LiMn}_2\text{O}_4$  was covered with fine particles. From XRD, EDAX, and TEM analyses, it was clarified that the fine particle on its surface was  $\text{LiCoO}_2$ .  $\text{LiCoO}_2$ -coated  $\text{LiMn}_2\text{O}_4$  showed an excellent cycle stability at  $65^\circ\text{C}$  compared to pure  $\text{LiMn}_2\text{O}_4$ . While pure  $\text{LiMn}_2\text{O}_4$  retained 70 % of the initial capacity after 100 cycles at the rate of 120 mAh/g at  $65^\circ\text{C}$ ,  $\text{LiCoO}_2$ -coated  $\text{LiMn}_2\text{O}_4$  retained approximately 92 % of the initial capacity. The improvement of cycle stability at  $65^\circ\text{C}$  is attributed to the reduction of Mn dissolution resulting from the suppression of the electrolyte decomposition induced by the catalytic activity of  $\text{LiMn}_2\text{O}_4$ . In the case of  $\text{LiNi}_{0.8}\text{Co}_{0.2}\text{O}_2$ -coated  $\text{LiMn}_2\text{O}_4$ , its high temperature cycle performance was also as excellent as that of  $\text{LiCoO}_2$ -coated  $\text{LiMn}_2\text{O}_4$ . Consequently, It is proposed that the surface encapsulation of  $\text{LiMn}_2\text{O}_4$  with fine  $\text{LiNi}_{1-x}\text{Co}_x\text{O}_2$  ( $x = 0.2$  and  $1$ ) particles improve its cycling stability at high temperature.