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휴대형 전자기기 및 이동 전원용  
Small Fuel Cell 기술

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장 혁

(삼성종합기술원)



# 휴대형 전자기기 및 이동 전원용 Small Fuel Cell 기술

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삼성종합기술원, 전기화학 Lab

산업은 물론 사회의 각 분야에 있어서 21 세기가 요구하는 기술분야는 기계, 화학, 재료, 전자등의 기간기술을 포함하여 정보화기술, 환경친화기술, 생명공학기술등 범위가 방대할 뿐 아니라 그 기대수준이 매우 높고, 개발 속도는 더욱 가속화 되고 있다. 여러가지 Social Needs 중 전기화학 기술이 담당 해야 할 기술 분야로는 무한청정 대체에너지 개발 과 휴대형 정보통신 단말기의 고기능화를 지원할 수 있는 이동 전원의 개발등일 것이고, Fuel Cell 은 이러한 분야의 일부를 해결할 수 있는 미래 주요 기술중의 하나이다.

본 발표에서는 상온작동, 고용량 및 고출력이 가능하며, 이차전지의 한계를 극복하는 차세대 휴대형 전자기기용 전원과 소형 이동전원은 물론 주택용 전원 및 무공해 자동차등 응용분야가 다양한 PEMFC (Proton Exchange Membrane Fuel Cell) 와 DMFC (Direct Methanol Fuel Cell) 기술을 소개하고자 한다. 특히 이에 필요한 요소 기술인 고출력밀도 촉매전극 및 MEA (Membrane & Electrode Assembly) 기술, 수소이온 전도막 기술, 그리고 Cell Pack 혹은 Stack 기술의 원리 및 최근 개발 동향에 대하여 언급하였다.

또한, 당 팀에서 개발한 Small Fuel Cell 기술 중 고효율 MEA 재료 및 공정 ( $0.3 \text{ W/cm}^2$ , Dry  $\text{H}_2/\text{Air}$ , 상온, 상압 ;  $0.1 \text{ W/cm}^2$ , 2M Methanol/Air, 80C, 상압), Nafion 대체 수소이온 전도막 재료 ( $0.1 \text{ S/cm}$ ), 그리고 소형 PEMFC Stack (40 W, 100 W, 200 W) 과 DMFC Cell Pack (600 mW) 기술을 소개하고, 선진기술 혹은 목표수준과 비교하여 앞으로 연구력을 집중해야 할 분야들을 소개함으로써 국내의 Small Fuel Cell 관련 기술의 발전 및 실용화를 도모하고자 한다.

# 휴대형 전자기기 및 이동 전원용 Small Fuel Cell 기술

삼성종합기술원  
장혁

Battery Symposium, Dec.1. 2000

## Social Needs

### ■ 전기화학기술이 담당 해야 할 21세기의 Social Needs

- ✓ 무한 청정 대체에너지
- ✓ 휴대형 정보통신 단말기의 고기능화를 지원할 수 있는  
고출력밀도, 고에너지밀도, 초소형화 및 충전시간 단축을 실현한 전원장치

### ■ Battery vs. Fuel Cell

#### Rechargeable Battery의 한계

제한된 사용시간  
긴 충전시간  
공해유발 발전소의 유지 필요



#### Fuel Cell로 해결

사용시간 연장 (Theo. 5 X)  
충전 불필요 (1분 이내 Refill)  
소형화 및 고출력 가능  
무한 청정에너지

## Research Objectives

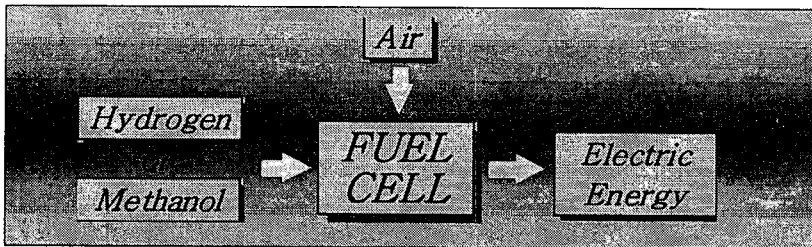
- 21세기 무공해 대체 에너지기술 확보
- Battery의 한계 (충전시간, 사용시간)를 극복하는 차세대 이동전원으로써 상온 작동, 고용량, 고출력 *Portable Fuel Cell*의 연구 개발



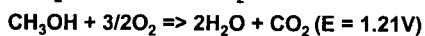
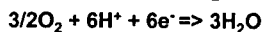
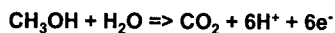
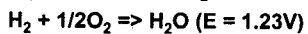
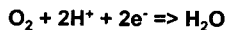
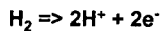
- 고출력 MEA (Membrane & Electrode Assembly) 개발
- 저가격 Proton Exchange Membrane 재료 개발
- PEMFC, DMFC Cell Pack / Stack 개발 및 실용화를 위한 요소기술 확보

## Technology Principle

### Fuel Cell Concept



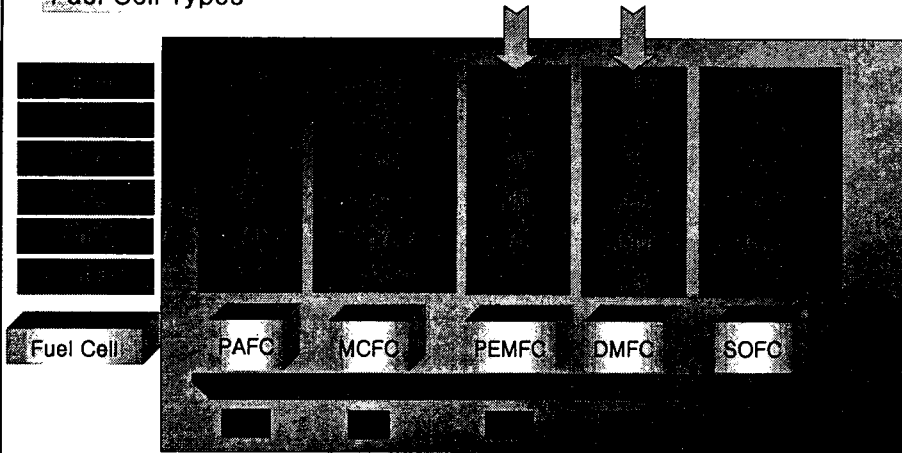
### Electrochemical Reactions



# Technology Principle



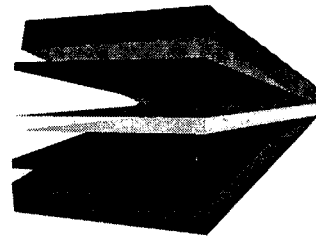
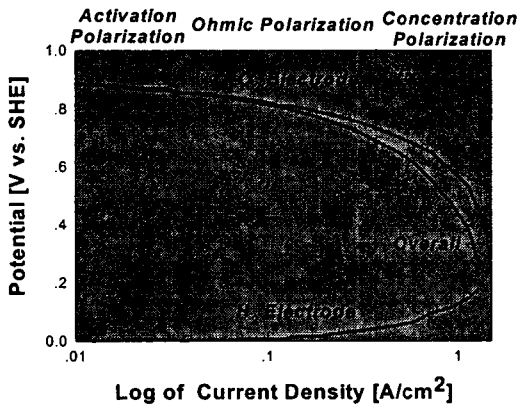
## Fuel Cell Types



# Technology Principle



## Polarization



MEA

## Applications (PEM & DMFC)

- Portable Electronic Devices (Note PC, Cellular Phone, PDA, Portable Power)
- Stationary Stand-alone Power (House Utilities)
- Fuel Cell Vehicle (ZEV or LEV)



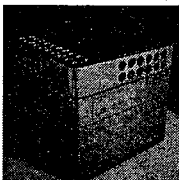
Ref. Samsung



Ref. Samsung



Ref. www.daisanalytic.com



Ref. www.ballard.com

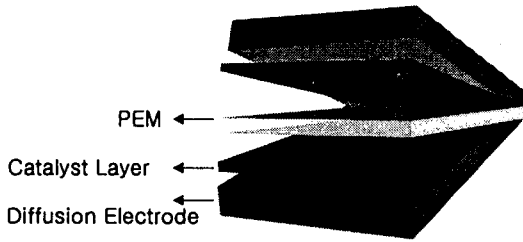


Ref. www.daimlerchrysler.de

## Technology Trend

Core Technology	Technical Issues
MEA	<ul style="list-style-type: none"> <li>· Catalyst Electrode Process</li> <li>· High Power Density MEA</li> <li>· Self Humidified (Water Management Electrode)</li> <li>· Air Breathing Diffusion Layer</li> <li>· CO Tolerant Catalyst</li> <li>· Catalyst Coating (Catalyzed Membrane)</li> </ul>
Membrane	<ul style="list-style-type: none"> <li>· High Proton Conductivity / Conductance</li> <li>· Chemical / Mechanical Stability</li> <li>· Low Cost</li> <li>· Low Methanol Cross-over</li> </ul>
Cell Pack	<ul style="list-style-type: none"> <li>· Thin &amp; Light Bipolar Plate</li> <li>· Cell Pack Design</li> <li>· Ambient Temp./Ambient Pressrue Operation</li> </ul>

고출력 밀도 MEA Process



- Catalyst Electrode
  - ✓ Spray Coating
  - ✓ Decal Process
  - ✓ Tape Casting...
- Catalyzed Membrane
  - ✓ Spray Coating
  - ✓ Vapor Deposition
- MEA Bonding

MEA Microstructure





## Technology Trend



### Why Pt

- $\text{H}_2$  dissociation  $\Delta G_{298} = 412 \text{ kJ/mol}$
- $\text{O}_2$  dissociation  $\Delta G_{298} = 468 \text{ kJ/mol}$
- $\text{H} + \text{O}_2$  or  $\text{H}_2 + \text{O}$  activation  $40 \text{ kJ/mol}$
- $2\text{Pt} + \text{H}_2 \rightarrow 2 \text{Pt-H}$  zero activation
- $2\text{Pt} + \text{O}_2 \rightarrow 2 \text{Pt-O}$  zero activation
- $1/2 \text{O}_2 + \text{H}_2 \rightarrow \text{H}_2\text{O}$   $\Delta G_{298} = -232 \text{ kJ/mol}$

### Why Ru

- activate water or form surface oxides at lower potential
- CO absorption bond is less strong on Pt-Ru
- Pt-Ru ensemble  
oxidative methanol dehydrogenation on Pt by oxygen-like species on Ru
- $\text{OH}_{\text{ads}}$  species on Pt-Ru pair sites

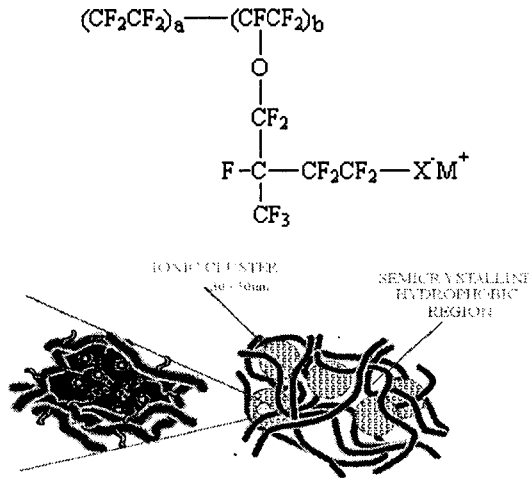
## Technology Trend



### Proton Exchange Membrane

- Strong functional radicals : high rapid exchange capacity
- Water dissociation of the polar sulfonic acid groups
  - ⇒ releases protons from  $\text{SO}_3^-$  groups
  - ⇒ hydrated
  - ⇒ protons move
- Ionic conductivity :  $\sim 0.1 \text{ s/cm}$
- Gore-Select :  $12 \mu\text{m}$ ,  $80 \text{ s/cm}$  at  $25^\circ\text{C}$ ,  $0.1 \text{ s/cm}$
- Nafion 117 :  $200 \mu\text{m}$ ,  $< 10 \text{ s/cm}$  at  $25^\circ\text{C}$ ,  $0.15 \text{ s/cm}$

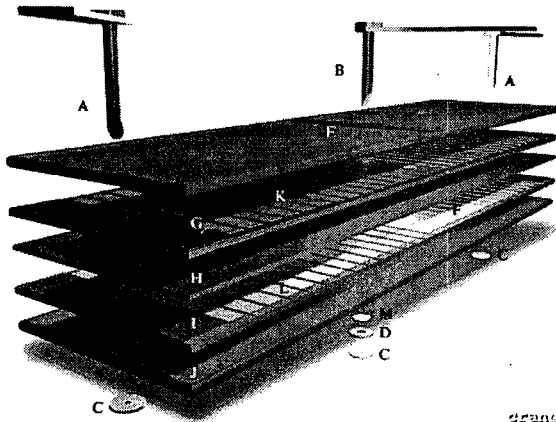
Nafion



Stack or Cell Pack

- JPL / USC / Giner DMFC  
25W ~ 200W for charger & cell phone (Composite membrane, PSSA/PVDF)
- H-Power PEMFC  
50W ~ 200W for message sign, military, telecommunication
- LANL / Motorola PEM / DMFC  
1W with dc/dc converter for cell phone
- Fraunhofer Inst. PEM / DMFC  
250mW ~ 30W for Note PC with dc/dc converter (Micromachined bipolar plate)
- MSI DMFC  
Nuclepore membrane with Pd layer for cell phone (300Wh/kg)
- SAIT PEM / DMFC  
1W ~ 200W for portable electronic devices

**Micro Fuel Cell™**



- A. Air Electrode Contact Rivet
- B. Fuel Needle
- C. Rivet Fold Out
- D. Upper Gasket Ring
- E. Lower Gasket Ring
- F. Through Contacts
- G. First Set of Fuel Cells
- H. Fuel Manifold
- I. Second Set of Fuel Cells
- J. Air Manifold
- K. Air Electrodes
- L. Fuel Electrodes
- M. Contact Washer

**MEA**

- Humid H<sub>2</sub>/O<sub>2</sub> 1Bar : 0.6 W/cm<sup>2</sup>
- Dry H<sub>2</sub>/Air 1Bar : 0.3 W/cm<sup>2</sup>
- 2M MeOH/Air 1Bar : 0.1 W/cm<sup>2</sup>
- PFSI Electrode & Catalyzed Membrane Process
- Water Management Electrode

**Membrane**

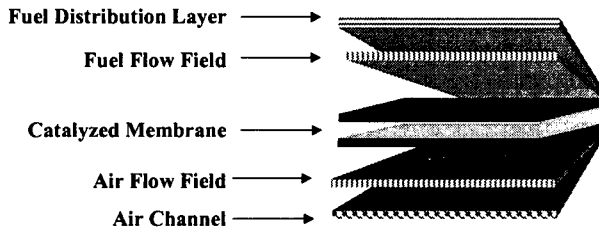
- Partially Fluorinated Membrane (Sulfonated PTFE)
- Composite Membrane

**Stack / Cell Pack**

- Light Weight / Thin Bipolar Plate
- Cell Unit Type Stack
- CAE Type Monopolar Pack
- 1W, 40W, 100W & 200W

## Technical Achievements at SAIT

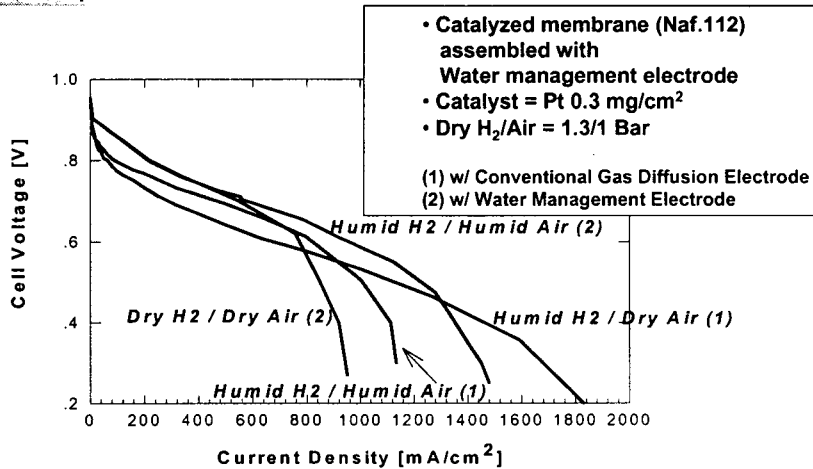
### MEA Process



MEA with Water Management Electrode

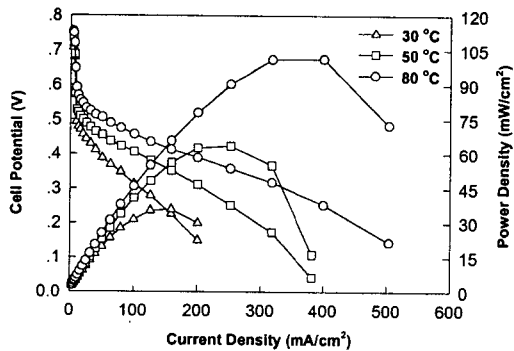
## Technical Achievements at SAIT

### H<sub>2</sub>-Air Operation (PEMFC)



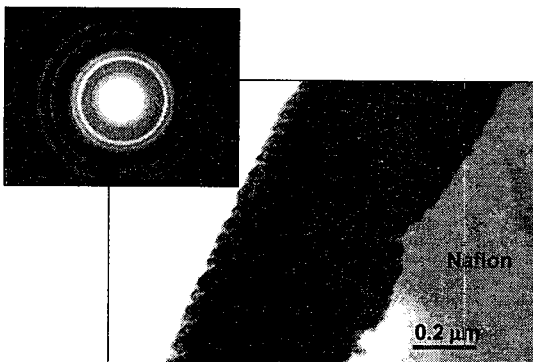
## Technical Achievements at SAIT

### Methanol -Air Operation (DMFC)



## Technical Achievements at SAIT

### TEM Microstructure (PtRu-C/Nafion)

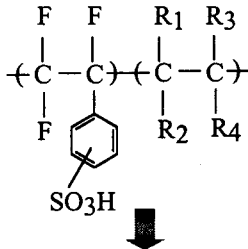


- Columnar growth with voids in between
- Rigid Contact
- Layer thickness : 0.5 μm (conventional : 20 ~ 30 μm)
- Nanocrystalline with 10 nm grain width
- Diffraction pattern shows fine crystalline
- Co-sputtering with Carbon retards the directional growth of PtRu
- Uniform distribution of PtRu and Carbon

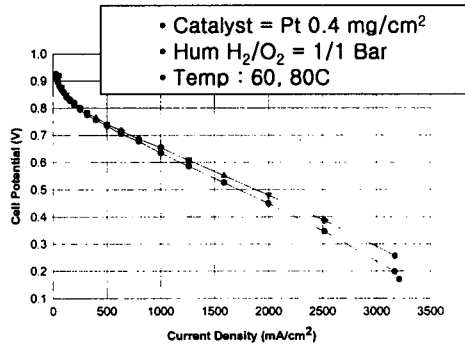
## Technical Achievements at SAIT



### Partially Fluorinated (Sulfonated PTFS) Membrane



- ✓ Low Cost Membrane
- ✓ High Protonic Conductivity
- ✓ Mechanical & Chemical Stability



## Technical Achievements at SAIT



### Composite Membrane

#### ■ Second Phase Dispersed Membrane

Matrix : Nafion

Dispersoid : Silica (3 wt.%, 15nm)

⇒ Thickness < 100 μm

Ionic Conductivity : 0.08 s/cm at R.T

#### ■ Reinforced Membrane

Matrix : Nafion / PTFS

Substrate : Microporous PTFE / PVDF

⇒ Thickness < 30 μm

Ionic Conductivity : 0.07 s/cm at R.T

- ✓ Water Absorption
- ✓ Mechanical Stability
- ✓ Thin & Higher Conductance
- ✓ Low Methanol Cross-over

## Technical Achievements at SAIT



### Light Weight Bipolar Plate

- Electrically Conducting Carbon Composite Laminate Plate
- Conductance :  $1850 \text{ s/cm}^2$
- Thickness :  $250 \sim 300 \mu\text{m}$

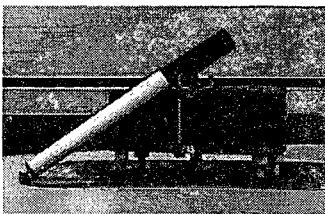


- ✓ Cost effective / Highly flexible / High conductance
- ✓ Gas tight at the hydrogen pressure of 1.4 Bar
- ✓ Attached gas flow field
  - ⇒ Anode : Parallel & orthogonal flow field
  - ⇒ Cathode : Thin carbon/plastic channel structure & flow field

## Technical Achievements at SAIT

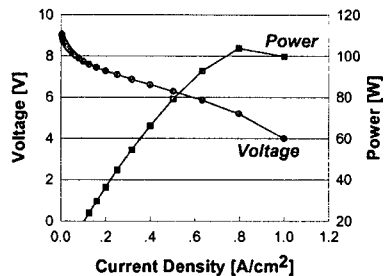


### Stack I



- ✓ Dry Fuel Operation
- ✓ Small & Light Stack

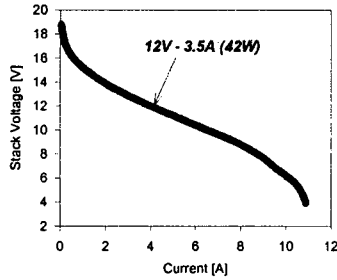
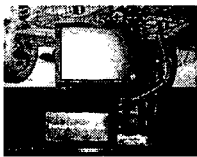
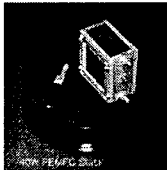
- Nominal Power : 90 W
- Operation : R.T ~ 70 °C, Dry H<sub>2</sub>/Air
- 150 g / 100 cc
- 600 W/kg, 900 W/L



## Technical Achievements at SAIT



### Stack II



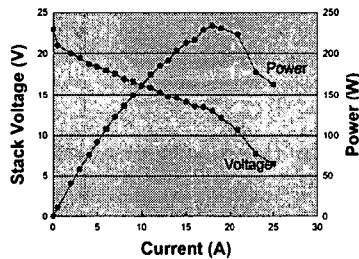
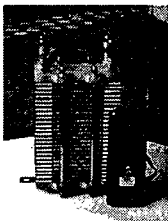
- Nominal Power: 40 W
- Energy Density: 140 Wh/kg (w/MH)
- Operation: 30°C, Dry H<sub>2</sub>/Air
- 90g / 90cc
- 440 W/kg, 440 W/L

- ✓ Ambient Temp. / Pressure Operation
- ✓ No Moving Part (Except for Mini Fan)
- ✓ Operating Note PC for 6 hrs w/o refueling

## Technical Achievements at SAIT



### Stack III



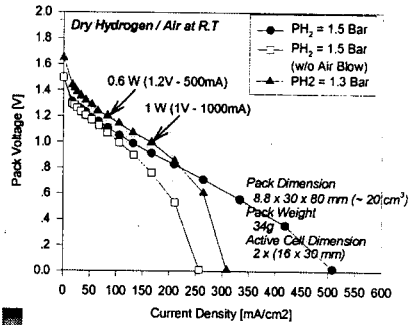
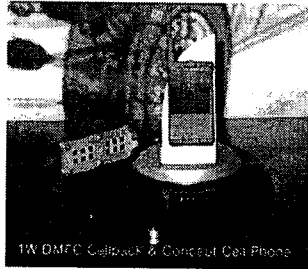
- Nominal Power : 200 W (12V)
- Max. Power : 250 ~ 300 W
- Operation: 40~50°C  
Dry H<sub>2</sub>/Air

- ✓ Portable Power Source
- ✓ Leisure or military application



## Technical Achievements at SAIT

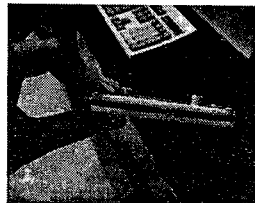
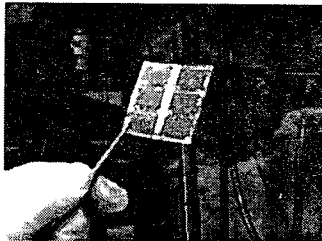
### Cell Pack I



- ✓ Monopolar Type 1W Cell Pack
- ✓ Concept Design of Cellular Phone Operated by Fuel Cell

## Technical Achievements at SAIT

### Cell Pack II



- Power Output: 2V - 40 mA
- Operation: Room Temp., 2M MeOH/Air
- 1.5g / 600 μm Thick

- ✓ Achievement of Basic Performance for 6 Cell Monopolar Cell Pack
- ✓ Under Development of Miniature Fuel Cell Technology

## Further Studies

- Methanol Oxidation Catalyst
- CO Tolerant Catalyst
- Oxygen Reduction Catalyst
- Catalyst Supporting Materials
- Catalyst Electrode Process
- Catalyzed Membrane Process
  - 고출력 밀도 MEA
- Monomer 합성 및 Polymer
- Nafion 대체 Ionomer
  - Methanol Cross-over Membrane
  - Composite Membrane 개발
  - Membrane Casting 공정
- Liquid Flow Field 재료 및 Design
- Gas Flow Field 재료 및 Design
- Methanol Storage 및 Flow Mechanism
- Circuit Design
- Sealing 기술
- Packaging 기술
- 연속 공정
- Cellular Phone용 Module
- Cellular Phone 탑재
- Cycle Test
- 고출력사용특성 개선

## Conclusion

(Energy Storage Device Technology Roadmap)

