

# SPECTROSCOPY AND DYNAMICS OF NEAR-INFRARED METASTABLE SPECIES IN SOLUTION PHASE

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Through the development of ultrasensitive near-infrared (NIR) detecting systems we have opened up a new territory both on NIR spectroscopy and dynamics in solution phase. We have explored the  $O_2$   $^1\Sigma_g^+ \rightarrow ^1\Delta_g$  (0,0) 1930 nm,  $^1\Sigma_g^+ \rightarrow ^3\Sigma_g^-$  (0,0) 765 nm and  $O_2$  dimol ( $^1\Delta_g$ )<sub>2</sub>  $\rightarrow$  ( $^3\Sigma_g^-$ )<sub>2</sub> (0,0) and (0,1) at 634 and 703 nm emission, respectively in  $CCl_4$ . Consequently, the photophysics of  $^1O_2$  dimol ( $^1\Delta_g$ )<sub>2</sub> and  $O_2$  ( $^1\Sigma_g^+$ ) states in solution has been studied in details. The quantum yield of  $O_2$   $^1\Sigma_g^+ \rightarrow ^3\Sigma_g^-$  765 nm emission was measured to be  $1.94 \times 10^{-7}$ , and consequently, the radiative decay rate of  $^1\Sigma_g^+ \rightarrow ^3\Sigma_g^-$  transition was deduced to be  $1.55 \pm 0.04$  s<sup>-1</sup>. By comparing the dimol ( $^1\Delta_g$ )<sub>2</sub>  $\rightarrow$  ( $^3\Sigma_g^-$ )<sub>2</sub> emission intensity with the tetra-tert-butylphthalocyanine delayed fluorescence induced by a two-step energy transfer from the  $O_2$  ( $^1\Delta_g$ ) state, a lower limit of the dissociation rate constant of the dimol was estimated to be  $(5.2 \pm 0.3) \times 10^{10}$  s<sup>-1</sup> in  $CCl_4$ . On the other hand, through the study of Nile Blue A delayed fluorescence induced by energy transfer from the  $^1O_2$  dimol, a radiative lifetime of  $(1.2 \pm 0.3) \times 10^3$  s<sup>-1</sup> for  $^1O_2$  dimol was deduced in  $CDCl_3$ .

In another approach, spectroscopy and dynamics of  $I(^2P_{1/2})$  in solution have been investigated via the iodine atom  $^2P_{1/2} \rightarrow ^2P_{3/2}$  emission in solution. The resulting emission spectra in the region of 1200-1600 nm reveal cooperative electronic-vibrational iodine-solvent transitions, furnishing insight into the atom-solvent interactions. The dynamics of energy-transfer between  $I(^2P_{1/2})$  and  $^1O_2$  in solution were first studied with detailed kinetics presented.