PHOTOCHEMICAL REACTION OF HALOARENE TETHERED TO ARENE WITH AMIDE BOND

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The photochemical behavior of 2-halo-N-pyridinylbenzamide and N-pyridinyl-2-Halobenzamide is studied. The photoreaction of 2-chloro-N-pyridinylbenzamide afforded photocyclized products, benzo[c]naphthyridinones in high yield, whereas the bromo analogs produced extensively photoreduced products, N-pyridinylbenzamides with minor Since photocyclization reaction of 2-chloro-*N*photocyclized product. the pyridinylbenzamide is retarded by the presence of oxygen and sensitized by the presence of a triplet sensitizer, acetone or acetophenone, a triplet state of the chloro analog is involved in the reaction. Since several radical intermediates are identified in the laser flash photolysis of 2-chloro-N-pyridinylbenzamide, a radical-mediated mechanism for the reaction is proposed: the triplet state (78 kcal/mol) of the chloro analog which is populated by the excitation of a distorted conformer having an intramolecular hydrogen bond interaction between a chlorine negative end and a hydrogen of the secondary amide bond and then by its intersystem crossing, undergoes a homolytic cleavage of the C-Cl bond to give phenyl and chlorine radicals. The intramolecular cyclization reaction of the phenyl radical with its close pyridinyl group proceeds to produce a conjugated 2,3-dihydropyridinyl radical and then the conjugated radical aromatizes to afford a cyclized product, benzo[c]naphthyridinone by ejecting a hydrogen radical. The photoreduction product can be formed by abstracting of hydrogen atom of the phenyl σ radical from the environment.

We recently and unexpectedly observed that a photoreaction of N-(2-chlorophenyl)-4-pyridinylcarboamide in the presence of base gave a 2-pyridinylbenzaxazole (25%), a product of an intramolecular photosubstitution of halogen of the aryl halide with the carbonyl oxygen of the amide bond (Scheme). Bromo analog afforded a 2-pyridinylbenzoxazole in 34% yield. The novelty of the reactions deserve special attention. In order to elucidate the reaction mechanism, kinetic and laser flash photolysis studies have been performed. The yield of 2-pyridinylbenzoxazole under dilute base was lower than that under concentrated base, even though the starting material underwent certain reaction further under dilute base.