

## S 2-2

**Intracellular pH Regulation in Cardiac Myocytes**Chae Hun Leem<sup>1\*</sup> and Richard D. Vaughan-Jones<sup>2</sup>Department of Physiology Ulsan University, College of Medicine<sup>1</sup>,  
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Intracellular pH( $\text{pH}_i$ ) regulation is very important to regulate the cellular functions of cardiac myocytes such as contractility, signal transduction, ion regulation, cell volume, and energy production etc. The resting  $\text{pH}_i$  was maintained at about 7.07 and strictly regulated within the range of  $\pm 0.1$ . Primary defence mechanism for  $\text{pH}_i$  change is the intracellular proton buffers. Intracellular proton buffering power consists of two different components such as intracellular intrinsic buffering power ( $\beta_i$ ) and  $\text{CO}_2$ -dependent buffering power( $\beta_{\text{CO}_2}$ ).  $\beta_i$  inversely varied with  $\text{pH}_i$  in a manner consistent with two principal intracellular buffers of differing concentration and  $\text{pK}$ . The value of  $\beta_{\text{CO}_2}$  monotonically increased with  $\text{pH}_i$  and  $\beta_{\text{CO}_2}$  was consistent with buffering in a cell fully open to  $\text{CO}_2$ . Recently, it was discovered four types of sarcolemmal carriers contributes the intracellular pH regulation in cardiac myocytes. Two of these are acid-equivalent extruders ( $\text{Na}^+$ - $\text{H}^+$  exchange, NHE and  $\text{Na}^+$ - $\text{HCO}_3^-$  symport, NBC) while two are acid-loaders ( $\text{Cl}^-$ - $\text{HCO}_3^-$  exchange, AE and  $\text{Cl}^-$ - $\text{OH}^-$  exchange, CHE). In 5%  $\text{CO}_2/\text{HCO}_3^-$ -buffered conditions, the acid extrusion on NHE and NBC increased as  $\text{pH}_i$  was reduced below the resting  $\text{pH}_i$ , with the greater increase occurring through NHE at  $\text{pH}_i < 6.90$ . The acid influx on AE and CHE was increased as  $\text{pH}_i$  was raised above the resting  $\text{pH}_i$ . NHE, with the greater increase occurring through AE at

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$\text{pH}_i > 7.15$ . At the resting  $\text{pH}_i$ , all four carriers were activated equally, albeit at a low rate (about 0.15 mM/min).  $\beta_{\text{CO}_2}$  was developed slowly because of low carbonic anhydrase activity and  $\text{CO}_2$  hydration rate increased only 3.7 times compared to natural hydration rate. The  $\text{pH}_i$  dependence of flux through the transporters, in combination with the  $\text{pH}_i$  and time dependence of intracellular buffering ( $\beta_i + \beta_{\text{CO}_2}$ ), was used to predict mathematically the recovery of  $\text{pH}_i$  following an intracellular acid or base load. Under several conditions the mathematical predictions compared well with experimental recordings, suggesting that the model of dual acid influx and acid efflux transporters is sufficient to account for  $\text{pH}_i$  regulation in the cardiac cell. Key properties of the  $\text{pH}_i$  control system will be discussed.