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## Reaction of Urnium Dioxide with CF<sub>4</sub>/O<sub>2</sub> r. f. Plasma

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## **Abstract**

Research on the etching reaction of  $UO_2$  in  $CF_2/O_2$  r. f. plasma is carried out at temperatures of up to 370°C under the total pressure of 0.3 Torr. The reaction rates are investigated as functions of  $CF_2/O_2$  ratio, plasma power, and substrate temperature. It is found that the highest etching rate is obtained at 20% $O_2$  mole fraction regardless of r. f. power and substrate temperature. The highest etching reaction rate at 370°C under 150W exceeds 1000monolayers/min., which is equivalent to 0.4  $\mu$ m/min. The mass spectrometry analysis result reveals that the major reaction product is uranium hexa-fluoride  $UF_6$ . Based on the experimental findings, dominant overall reaction of uranium dioxide in  $CF_2/O_2$  plasma is determined:

$$8UO_2 + 12CF_4 + 3O_2 = 8UF_6 + 12CO_{2-x}$$

where CO<sub>2-x</sub> represents the undetermined mix of CO<sub>2</sub> and CO.