

The Analysis of N Component in Thin Oxide Film Thermally Grown by NH₃ Oxidation

Young-Cho Kim and Chul-Ju Kim

Dept. of Electronic Engineering, Seoul City Univ.,
90 Jeon Nong-Dong, Dong Dae Moon-Ku, Seoul 130-743, Korea

ABSTRACT

The depth profiles of the as-grown and the annealed NH₃ oxide film in NH₃(7.5%)/N₂ ambient at 450°C are analyzed. This annealing in the ambient of mixed gases removes the small quantities of N component from the NH₃ oxide film. In AES analysis, the NH₃ oxidation shows the exact stoichiometry of SiO₂ and a sharp sloop at SiO₂/Si interface

I. Introduction

The new thermal oxidation methods, which is done by adding NH₃ gases to pure O₂ gases and proposed to promote the material quality of oxide film and the SiO₂/Si interface¹⁾, practically enable fabrication process due to its simplicity. As a result so far, it is known that reaction equation of NH₃ oxidation is $4\text{NH}_3 + 3\text{O}_2 \rightarrow 2\text{N}_2 + 6\text{H}_2\text{O}$ ²⁾, and the quantities of oxidizing species³⁾, O₂ and H₂O, are controlled by concentration of NH₃ gases⁴⁾ added during oxidation. But none of the other material analyses for the oxide layer thermally grown by this new method are studied.

In this study, we obtained the depth profile of the as-grown NH₃ oxide film and the annealed NH₃ oxide film in NH₃(7.5%)/N₂ ambient at 450°C. We analyzed the atomic concentration of this film for the stoichiometry of SiO₂ film and SiO₂/Si interface states using AES(Auger Electron Spectroscopy).

II. Experimentation and Results

Fig.1 shows the depth profile using SIMS(Secondary Ion Mass Spectroscopy) for the oxide film grown by NH₃ oxidation method at 1100°C for the wafer of (100), p-type, 6-7 Ω·cm. In this figure the quantity of N components existing in the oxide layer and Si substrate is very small compared to that of O₂ and Si component to the 4~5 orders.

Fig.2 shows the depth profile using SIMS for the NH₃ oxide film annealed at 450°C in the same atmosphere of Fig.1. As shown in this figure, N components existed in oxide film are almost removed. We consider that the N components are substituted to the H components decomposed from the NH₃ gases.

Fig.3 is a qualitative analysis of the NH₃ oxide film using AES under the same condition of Fig.1. In this figure the oxide film grown by NH₃ oxidation method is certainly composed of SiO₂ stoichiometry, considering the Si:O ratio of 1:2, and the peak

of N is very small like carbon as a contamination. Observing the slope at Si-SiO₂ interface, its sharp slope and intersecting of Si and O in the ratio of 1:1 show the fixed surface charge is minimized.

III. Conclusion

In this study N components can be ignored by very small quantities, and the very small N components are removed by an annealing of NH₃(7.5%)/N₂ ambient at 450°C observing the depth profile of SIMS. In qualitative analysis using AES, NH₃ oxidation generates exact stoichiometry of SiO₂ in NH₃ oxide film and sharp slope to minimize the surface charge at the interface state.

Reference

- 1) Young-Cho Kim and Chul-Ju Kim, ICEC'92, 1199-102(1992)
- 2) Young-Cho Kim and Chul-Ju Kim, Jpn. Soc. Appl. Phys., 53, 636(1992)
- 3) Young-Cho Kim and Chul-Ju Kim, JTC-CSCC'92, 214-217(1992)
- 4) Y. C. Tseng and K. Shono, Jpn. J. Appl. Phys., 28, L329-L331(1989)

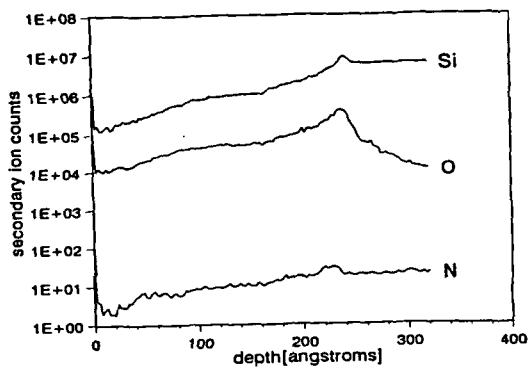


Fig. 1. The depth profile of the as-grown NH₃ oxide film using SIMS

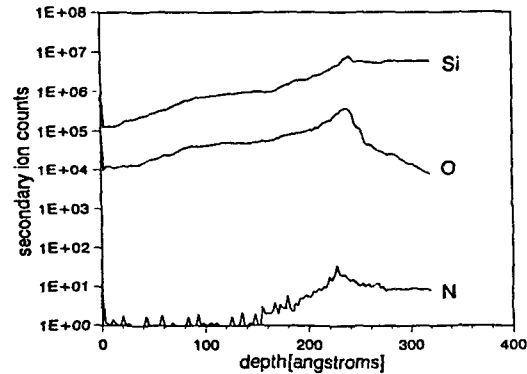


Fig. 2. The depth profile of NH₃ oxide film annealed at 450°C in NH₃(7.5%)/N₂ ambient.

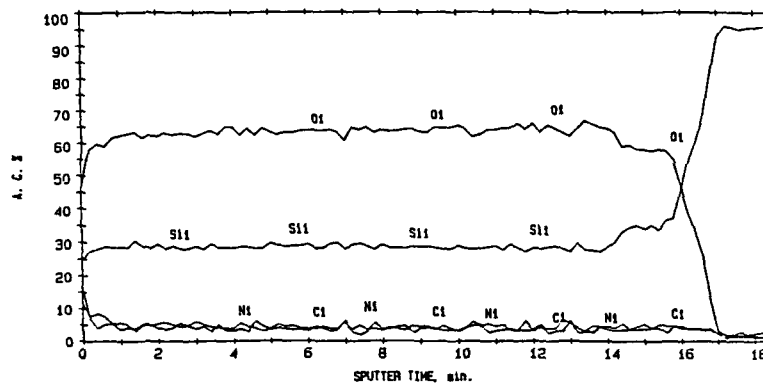


Fig. 3. The normalized atomic concentration by AES.